

M0140 Salt Spray Test

The Standard Salt-Spray Test--Is it a Valid Acceptance Test?Corrosion Tests and StandardsNew Methods for Corrosion Testing of Aluminum AlloysCorrosion Testing and EvaluationCorrosion Tests in Artificial Atmospheres. Salt Spray TestsCyclic Cabinet Corrosion TestingThe Salt Spray Test and Its Use in Ranking Stainless SteelsCorrosionSalt-Spray TestingMaterials Evaluation Using Wet-Dry Mixed Salt-Spray TestsSalt Spray Test Vs. Field Results for Coated SamplesCorrosion Tests in Artificial AtmospheresSALT SPRAY TESTING AND EVALUATION OF FASTENER FINISHESSalt Spray TestMethod for Salt Spray Corrosion Tests in Artificial AtmospheresEngineering Materials and Processing MethodsStandard Practice for Modified Salt Spray (fog) TestingASTM Special Technical PublicationSalt Spray Test Vs. Field Results for Coated SamplesProceedings of Educational Sessions of the ... Annual Convention ... A. Mendizza Robert Baboian Vinod S. Agarwala Robert Baboian British Standards Institute Staff Gardner S. Haynes Joseph R. Davis J. J. Bowman JB. Johnson Amy Forsgren Standards South Africa USCAR. W. H. Mutchler British Standards Institute Staff American Society for Testing and Materials Amy Forsgren American Electroplaters' Society. Convention

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the salt spray test was originally developed some 40 odd years ago 1 for the purpose of quickly evaluating the resistance of plated coatings to attack by marine atmospheres through the years an ever increasing use has been made of this test for the purpose of controlling quality of electroplated articles today the salt spray test is employed not only to evaluate the quality of all types of electroplated coatings both of the cathodic and of the anodic type but is also frequently used for testing completely assembled equipment many specifications contain requirements based on performance in the salt spray compliance with which is rigidly enforced the endpoint requirements will vary depending on the type of coating tested or on the combination of dissimilar metals used in the case of electropositive coatings the endpoint is usually specified in terms of a permissible number of corrosion spots per unit area that will develop in a given time electronegative coatings such as zinc and cadmium on steel are evaluated by their ability to prevent rust formation notwithstanding its widespread use the salt spray test has frequently been subject to criticism for being unrealistic and misused 2 3 4 its role as a forecaster of service performance has been challenged and failure to correlate salt spray test results with performance to actual atmospheric environments have been reported in the literature 5 6 7 inability to obtain reproducible results both within any one test box as well as amongst several boxes has also been unofficially reported by many users of this test although its shortcomings are thus generally recognized the salt spray test continues to form a part of many finish specifications both in industry and government agencies the lack of a more suitable replacement is the most frequent reason given for the retention of the salt spray test as a standard test the american society for testing materials has standardized a salt spray testing procedure with its method b 117 the uniformity of conditions within the test chamber operated in accordance with this test procedure is controlled primarily by three physical parameters rate of salt fog precipitation concentration of collected salt solution and ambient temperature rather wide limits are allowed for the precipitation rate the concentration of collected solution and

the temperature are kept within much narrower limits in addition the chemical purity of the salt and the ph of the salt solution are also controlled no provisions are made for calibrating the salt spray test chamber against some standard government specifications on the salt spray are in the main similar to those covered by the astm method

evaluates the usefulness of the current standards on exfoliation and corrosion testing of aluminum alloys and their applicability to new requirements and advanced alloys the 13 papers from an international symposium in san francisco may 1990 discuss whether the existing standards should be revis

thirty papers provide information on the magnitude of corrosion damage and how testing and evaluation techniques assist in minimizing failures new developments in computer aided evaluations are highlighted along with advances in electrochemical techniques also covered are measurements in soil wat

salt spray tests accelerated corrosion tests corrosion tests artificial weathering tests environmental testing metals alloys metal coatings protective coatings coatings decorative coatings organic coatings conversion coating electrochemical coating chemical resistance tests acetic acid test equipment

as the title suggests this is an introductory book covering the basics of corrosion it is intended primarily for professionals who are not corrosion experts but may also be useful as a quick reference for corrosion engineers included in the 12 chapters are discussions of the physical principles and characteristics of corrosion help in recognizing and preventing corrosion and techniques for diagnosing corrosion failures

although theoretical studies afford valuable information simple practical tests are of marked assistance in predicting the resistance of materials to specific corrosive conditions one of the oldest of these is the salt spray the authors believe much of its unpopularity results from uncontrolled variables and the lack of standardized procedures and they present data to show that the test is both useful and reliable when properly administered the factors which they believe should be controlled are briefly discussed and standard conditions for each factor are suggested however the paper has

been written to stimulate discussion of the test and promote more universal agreement on the testing methods used and is not intended as a standard practice

accelerated methods for corrosion testing of materials based on the use of salt sprays have been standardized since the early 1930s however despite the many significant advances in the mechanistic understanding of atmospheric corrosion phenomena since those days most changes in accelerated test standards have been implemented to reduce the time of the tests and thus their costs by increasing their severity thus although laboratory accelerated test methods have played an important role in the assessment of materials performance in general they have not been applied by virtue of their ability to simulate and enhance natural weathering in particular wet dry cycles simulate the natural wetting and drying which occurs in practice and under many conditions can aid in the formation of naturally occurring corrosion product films which frequently play a role in the kinetics of metal corrosion in addition the use of a relatively dilute mixed salt spray in comparison with the more commonly used solutions based on 5 nacl e g astm b 117 method for salt spray fog testing can result in corrosion morphologies and behavior which are more representative of natural conditions this paper describes a wet dry mixed salt spray test using solutions containing mixtures of $\text{nh}_4\text{ }_2\text{so}_4$ and nacl with a 1 h wet 1 h dry cycle the basis and reasons for using the test are discussed in depth with the emphasis on illustrating the improvements in realism which are attained using this procedure rather than a standard continuous nacl test in addition the results from accelerated corrosion testing of aluminum are presented to illustrate the main features of the test finally possible developments and improvements in the salt spray methods including the use of solutions resembling artificial rainwater are discussed

this procedure is used to test and evaluate the resistance of fastener surface finishes to laboratory salt spray testing

salt spray tests accelerated corrosion tests corrosion tests acetic acid testing conditions test equipment metals specimen preparation alloys test specimens

issues for 1929 include section contents noted 1929 1939 called metallurgical abstracts jan 1940 sept 1945 called engineering digest oct 1945 called materials methods digest

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