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Annual Review of Physical Chemistry Applications of MO Theory in Organic Chemistry Mansoor Niaz Darrell D. Ebbing Professor  
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this book explores the relationship between the content of chemistry education and the history and philosophy of science hps framework that underlies such education it discusses the need to present an image that reflects how chemistry developed and progresses it proposes that chemistry should be taught the way it is practiced by chemists as a human enterprise at the interface of scientific practice and hps finally it sets out to convince teachers to go beyond the traditional classroom practice and explore new teaching strategies the importance of hps has been recognized for the science curriculum since the middle of the 20th century the need for teaching chemistry within a historical context is not difficult to understand as hps is not far below the surface in any science classroom a review of the literature shows that the traditional chemistry classroom curricula and textbooks while dealing with concepts such as law theory model explanation hypothesis observation evidence and idealization generally ignore elements of the history and philosophy of science this book proposes that the conceptual understanding of chemistry requires knowledge and understanding of the history and philosophy of science professor niaz s book is most welcome coming at a time when there is an urgently felt need to upgrade the teaching of science the book is a huge aid for adding to the usual way presenting science as a series of mere facts also the necessary mandate to show how science is done and how science through its history and philosophy is part of the cultural development of humanity gerald holton mallinckrodt professor of physics professor of history of science harvard university in this stimulating and sophisticated blend of history of chemistry philosophy of science and science pedagogy professor mansoor niaz has succeeded in offering a promising new approach to the teaching of fundamental ideas in chemistry historians and philosophers of chemistry and above all chemistry teachers will find this book full of valuable and highly usable new ideas alan rocke case western reserve university this book artfully connects chemistry and chemistry education to the human context in which chemical science is practiced and the historical and philosophical background that illuminates that practice mansoor niaz deftly weaves together historical episodes in the quest for scientific knowledge with the psychology of learning and philosophical reflections on the nature of scientific knowledge and method the result is a compelling case for historically and philosophically informed science education highly recommended harvey siegel university of miami books that analyze the philosophy and history of science in chemistry are quite rare chemistry education and contributions from history and philosophy of science by mansoor niaz is one of the rare books on the history and philosophy of chemistry and their importance in teaching this science the book goes through all the main concepts of chemistry and analyzes the historical and philosophical developments as well as their reflections in

textbooks closest to my heart is chapter 6 which is devoted to the chemical bond the glue that holds together all matter in our earth the chapter emphasizes the revolutionary impact of the concept of the covalent bond on the chemical community and the great novelty of the idea that was conceived 11 years before quantum mechanics was able to offer the mechanism of electron pairing and covalent bonding the author goes then to describe the emergence of two rival theories that explained the nature of the chemical bond in terms of quantum mechanics these are valence bond vb and molecular orbital mo theories he emphasizes the importance of having rival theories and interpretations in science and its advancement he further argues that this vb mo rivalry is still alive and together the two conceptual frames serve as the tool kit for thinking and doing chemistry in creative manners the author surveys chemistry textbooks in the light of the how the books preserve or not the balance between the two theories in describing various chemical phenomena this talmudic approach of conceptual tension is a universal characteristic of any branch of evolving wisdom as such mansoor s book would be of great utility for chemistry teachers to examine how can they become more effective teachers by recognizing the importance of conceptual tension sason shaik saeree k and louis p fiedler chair in chemistry director the lise meitner minerva center for computational quantum chemistry the hebrew university of jerusalem israel

the sci fi film the matrix introduces a fascinating premise where humans function as energy sources for an advanced machine society in this fictional world human bodies are maintained in a state of suspended animation while their minds exist in a virtual reality allowing machines to extract their bioelectric thermal and kinetic energy this article investigates the scientific feasibility of utilizing humans as a power source by applying thermodynamic principles according to the first law of thermodynamics the energy required to sustain human life would result in a net energy loss for the machines the second law indicates that the system s entropy would rise rendering it an inefficient energy strategy furthermore the energy output of a human body even if fully utilized would be inadequate to meet the machines energy demands more efficient alternatives for the machines would include other biological power sources and energy harvesting techniques such as solar or nuclear power the article concludes that while the concept of human batteries serves as an engaging storytelling element it is not a scientifically viable solution for the machines energy requirements the machines choice to preserve human life may be motivated by other factors such as leveraging their collective cognitive abilities for computational purposes or adhering to an ethical code that prohibits the complete annihilation of humanity this investigation aims to fill the gap by providing a detailed thermodynamic analysis of the energy expenditure required to sustain human life in a suspended animation state and the inefficiency of this system as an energy source for machines a facet previously unexplored by elucidating the thermodynamic constraints of human based energy sources this study not only challenges a popular sci fi narrative but also enriches our understanding of bioenergetic processes and their implications for future energy harvesting technologies

this book explores the evolving nature of objectivity in the history of science and its implications for science education it is generally considered that objectivity certainty truth universality the scientific method and the accumulation of experimental data characterize both science and science education such universal values associated with science may be challenged while studying controversies in their original historical context the scientific enterprise is not characterized by objectivity or the scientific method but rather controversies alternative interpretations of data ambiguity and uncertainty although objectivity is not synonymous with truth or certainty it has eclipsed other epistemic virtues and to be objective is often used as a synonym for scientific recent scholarship in history and philosophy of science has shown that it is not the experimental data baconian orgy of quantification but rather the diversity plurality in a scientific discipline that contributes toward understanding objectivity history of science shows that objectivity and subjectivity can be considered as the two poles of a continuum and this dualism leads to a conflict in understanding the evolving nature of objectivity the history of objectivity is nothing less than the history of science itself and the evolving and varying forms of objectivity does not mean that one replaced the other in a sequence but rather each form supplements the others this book is remarkable for its insistence that the philosophy of science and in particular that discipline's analysis of objectivity as the supposed hallmark of the scientific method is of direct value to teachers of science meticulously yet in a most readable way mansoor niaz looks at the way objectivity has been dealt with over the years in influential educational journals and in textbooks it's fascinating how certain perspectives fade while basic questions show no sign of going away there are few books that take both philosophy and education seriously this one does roald hoffmann cornell university chemist writer and nobel laureate in chemistry

the book is a simple to understand low priced chemistry text with many worked out examples in topics which students have the most problems it is intended to serve as a guide to the teaching of chemistry on the one hand and for the student's own understanding of the principles in the areas they feel deficient the material is presented in very simple english and several worked out calculations in problematic areas have been included in addition the presentation is like the teacher is talking to the student and consequently the student should be at ease in understanding the chemistry concepts and the examples given should bring them closer to liking the subject

whether you're an avid student or an inquisitive learner the chemistry connection from atoms to applications is your key to unlocking the amazing world of chemistry this book breaks down the basic components of matter atoms molecules and chemical reactions into clear explanations simplifying complicated ideas this book makes the connections demonstrating how chemistry affects everything around us from the smallest particles to the most significant applications in daily life you will teach about the amazing mechanisms that underpin everything in our world including the food we consume the technologies we use and even the surrounding natural

beauty through lucid illustrations meaningful comparisons and useful advice the chemistry connection makes science approachable and interesting for all readers this book provides a thorough exploration of the fundamentals of chemistry and its practical applications making it ideal for anybody wishing to brush up on their knowledge develop a better understanding of the topic or just quench their curiosity explore and learn how atom relates to your surroundings

this book provides notes for basic laboratory experiments in qualitative analysis of cations the book introduces readers to basic methods and laboratory safety subsequent chapters cover six groups of cations each chapter explains important details that are required to understand how a particular analytical method works for detecting cations in samples starting from sedimentation and ending with the identification key features simple reader friendly format introductory notes and summary covers several groups of metals appendix for handy reference with tables and references this is a useful textbook for early chemistry students and teachers as it equips the readers with sufficient information required to analyze chemical samples and deduce the presence of specific cations as part of laboratory coursework

the tenth edition of this market leading text has been substantially revised to meet the rapidly changing instructional demands of general chemistry international edition professors known for its carefully developed thoroughly integrated step by step approach to problem solving general chemistry international edition helps students master quantitative skills and build a lasting conceptual understanding of key chemical concepts the tenth edition retains this hallmark approach and builds upon the conceptual focus through key new features and revisions

chemistry seeks to provide qualitative and quantitative explanations for the observed behaviour of elements and their compounds doing so involves making use of three types of representation the macro the empirical properties of substances the sub micro the natures of the entities giving rise to those properties and the symbolic the number of entities involved in any changes that take place although understanding this triplet relationship is a key aspect of chemical education there is considerable evidence that students find great difficulty in achieving mastery of the ideas involved in bringing together the work of leading chemistry educators who are researching the triplet relationship at the secondary and university levels the book discusses the learning involved the problems that students encounter and successful approaches to teaching based on the reported research the editors argue for a coherent model for understanding the triplet relationship in chemical education

an outline of the basic concepts of chemistry includes discussions of scientific notation atomic structure chemical bonding and the

periodic table

provides abstracts and review articles on topics in physical chemistry

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