

Flinn Scientific Factors Affecting Reaction Rates Answers

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solvent effects on reaction rates and mechanisms is a title that will conjure up visions of different things to different investigators in the field of reaction kinetics the physical chemist will envision the effects on rates of reactions of dielectric constant viscosity internal cohesion and external

pressure as these are influenced by the solvent the physical organic chemist will perhaps call to mind acidity basicity hydrogen bonding structure effects electro negativity and solvating ability as related to the solvent the strictly organic chemist may simply think in terms of a medium in which reactants can be made to form products merely because of solubility relations and his choice of solvent may depend on the ease of obtaining in a reasonable length of time a relatively pure product by extraction or other procedures and in fact the topic includes all these and much more some of the phenomena are merely recorded as experimental observations some factors are subject to theoretical explanation but even when theoretically explained or mathematically formulated they may not be sufficiently dominant to justify the application of the theory other effects may not be subject to theoretical explanations but may be included in correlations that are widely applicable the explanation of some effects may be purely conjectural but at least give some satisfaction to the seekers of the answer to the question why the material presented in the following pages will run the gamut of all the above possibilities

the study guide reflects the unique problem solving approach taken by the chemical principles text the new edition of the study guide includes many new worked out examples

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