

Concise Inorganic Chemistry U K

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this is a textbook for advanced undergraduate inorganic chemistry courses covering elementary inorganic reaction chemistry through to more advanced inorganic theories and topics the approach integrates bioinorganic environmental geological and medicinal material into each chapter and there is a refreshing empirical approach to problems in which the text emphasizes observations before moving onto theoretical models there are worked examples and solutions in each chapter combined with chapter ending study objectives 40 70 exercises per chapter and experiments for discovery based learning

selected topics in inorganic chemistry is a comprehensive textbook discussing theoretical aspects of inorganic chemistry uniqueness of the book lies in treatment of all fundamental concepts such as structure of atom chemical bonding inner transition elements and coordination

chemistry with a modern approach illustration of text with relevant line diagrams and tabular presentation of data makes understanding of concepts lucid and simple the book is designed for b sc honours and m sc students

house s descriptive inorganic chemistry third edition provides thoroughly updated coverage of the synthesis reactions and properties of elements and inorganic compounds ideal for the one semester acs recommended sophomore or junior level course in descriptive inorganic chemistry this resource offers a readable and engaging survey of the broad spectrum of topics that deal with the preparation properties and use of inorganic materials using rich graphics to enhance content and maximize learning the book covers the chemical behavior of the elements acid base chemistry coordination chemistry organometallic compounds and numerous other topics to provide a coherent treatment of the field the book pays special attention to key subjects such as chemical bonding and buckminster fullerenes and includes new and expanded coverage of active areas of research such as bioinorganic chemistry green chemistry redox chemistry nanostructures and more highlights the earth s crust as the source of most inorganic compounds and explains the transformations of those compounds into useful products provides a coherent treatment of the field covering the chemical behavior of the elements acid base chemistry coordination chemistry and organometallic compounds connects key topics to real world industrial applications such as in the area of nanostructures includes expanded coverage on bioinorganic chemistry green chemistry redox chemistry superacids catalysis and other areas of recent development

part a overviews of biological inorganic chemistry 1 bioinorganic chemistry and the biogeochemical cycles 2 metal ions and proteins binding stability and folding 3 special cofactors and metal clusters 4 transport and storage of metal ions in biology 5 biominerals and biomineralization 6 metals in medicine part b metal ion containing biological systems 1 metal ion transport and storage 2 hydrolytic chemistry 3 electron transfer respiration and photosynthesis 4 oxygen metabolism 5 hydrogen carbon and sulfur metabolism 6 metalloenzymes with radical intermediates 7 metal ion receptors and signaling cell biology biochemistry and evolution tutorial i fundamentals of coordination chemistry tutorial ii

this textbook provides essential information for students of inorganic chemistry or for chemists pursuing self study the presentation of topics is made with an effort to be clear and concise so that the book is portable and user friendly inorganic chemistry 2e is divided into five major themes structure condensed phases solution chemistry main group and coordination compounds with several chapters in each there is a logical progression from atomic structure to molecular structure to properties of substances based on molecular structures to behavior of solids etc the author emphasizes fundamental principles including molecular structure acid base chemistry coordination chemistry ligand field theory and solid state chemistry and presents topics in a clear concise manner there is a reinforcement of basic principles throughout the book for example the hard soft interaction principle is used to explain hydrogen bond strengths strengths of acids and bases stability of coordination compounds etc the book contains a balance of topics in theoretical and descriptive chemistry new to this edition new and improved illustrations including symmetry and 3d molecular orbital representation expanded coverage of spectroscopy instrumental techniques organometallic and bio inorganic chemistry more in text worked out examples to encourage active learning and to prepare students for their exams concise coverage maximizes student understanding and minimizes the inclusion of details students are unlikely to use discussion of elements begins with survey chapters focused on the main groups while later chapters cover the

elements in greater detail each chapter opens with narrative introductions and includes figures tables and end of chapter problem sets

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