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the first law of thermodynamics $\Delta U = Q - W$ ΔU is the change in internal energy, Q is the heat added to the system, and W is the work done by the system.

thermodynamics

borgnakke c and sonntag r e 2014 fundamentals of thermodynamics 8th edition singapore wiley

ppt molecular thermodynamics of fluid
 phase equilibria

thermodynamics1854 on the dynamical theory of heat 1
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spontaneity

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