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the first law of thermodynamics $\Delta U = Q - W$ where ΔU is the change in internal energy, Q is the heat added to the system, and W is the work done by the system.

borgnakke c and sonntag r e 2014 fundamentals of thermodynamics 8th edition singapore wiley

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in 2011, the first law of thermodynamics was revised to include the concept of entropy.

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