

Crc Handbook Of Laboratory Safety

Handbook of Laboratory SafetyHandbook of Laboratory Health and Safety
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Norman V. Steere S. B. Pal Conrad L. Stanitski Michael H. Gabb Anne M. Van Leeuwen
David S. Jacobs V.H. TALIB Bernard D. Bolas Norman L. Parr (Ed) Josehine M.
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Williams & Wilkins Allied Laboratories Drew Provan Anne M. Van Leeuwen Norman
Lawrence PARR Gary S. Coyne

a concise and handy guide to the numerous recipes for chemical solutions used in laboratories in each chapter preparations of one particular use or related uses are grouped alphabetically where appropriate the use of the solutions are stated and cross reference made should meet most of the everyday requirements of chemistry physics biology and engineering laboratories contents foreword abbreviations authors note 1 solutions basic definitions molar and normal solutions standard solutions the purity of chemical substances 2 solutions handling techniques clean apparatus measuring or graduated apparatus the pipette the burette making a solution of approximate concentration making standard solutions by weighing standardization of solutions by titration cleaning solutions 3 solutions for titrations primary standards 1 standardization of strong acids 2 standardization of alkaline solutions 3 standardization of oxidizing agents 4 precipitation reactions 5 iodine titrations acids and alkalis solutions for redox reactions 1 oxidizing agents oxidants 2 reducing agents precipitation titrations miscellaneous titration solutions 4 bench solutions acids alkalis other inorganic reagents 5 indicators acid base or ph indicators screened indicators mixed indicators water soluble indicators other ph indicators luminescent indicators universal indicators buffer solutions indicators for precipitation titrations adsorption indicators starch indicator for iodine titration indicators for redox oxidation reduction reactions titrimetric or volumetric indicators indicators for edta titrations 6 organic reagents and others used in qualitative analysis 7 reagents used in organic chemistry 8 biochemical solutions and reagents 9 solutions in histology 10 physiological salines and culture solutions physiological salines animal plant culture solutions 11 miscellaneous solutions solutions for making indicator papers electrolyte solutions for cells and electrolysis appendix maximum tolerances in graduated glassware mathematical tables atomic weight table simple first aid procedures bibliography index

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